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-Completing chemical

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Aluminum reacts with

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oxygen to produce aluminum oxide as follows: $4\text{Al} + 3\text{O}_2 \rightarrow 2\text{Al}_2\text{O}_3$

a. If you use 2.3 moles of Al, how many moles of Al_2O_3 can you make?

b. If you want 3.9 moles of Al_2O_3 , how many moles of O_2 are needed?

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a. If you use 2.3 moles of Al, how many moles of Al_2O_3 can you make?
b. If you want 3.9 moles of Al_2O_3 Chemistry

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Mole, Grams to

Grams, Mole Ratio

Practice Problems

This chemistry video

tutorial provides a

basic introduction into

stoichiometry. It

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The Stoichiometry

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EQUATIONS (pages 353–358) This section explains how to calculate the amount of reactants required or product formed in a nonchemical process. It teaches you how to interpret chemical equations in terms of interacting moles, representative particles, masses, and gas volume at

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