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(MSBTE) Practical
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CHEMISTRY IIA

INTRODUCTION The
practical part of the
course covers
experiments on the
synthesis of organic
compounds and drugs
using different
analysis and
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techniques and includes synthesis of chiral drugs and the techniques used in the separation of enantiomers.

EQUIPMENT AND INSTRUMENTS

Analytical Balances

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Manual produce pure

biologically active

molecules to be

formulated as

medicines. An in

depth understanding

of organic chemistry

is core to the study of

this subject as it plays

a key role in many of

the processes utilised

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INTRODUCTION : #1

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PROCEDURE:

STEP-1: In a 2 lit RBF

fitted with a double

surface reflux

condenser place,

11.5gm (0.5gm atom)

of clean sodium and it

is dissolved in 250ml

of absolute alcohol.

To this solution add

80gm (0.5mole of

ethyl malonate)

followed by 30gm

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(0.5mole of dry urea)
dissolved in 250ml of
ethanol.

Laboratory

Medicinal chemistry

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Practical Inorganic II
Chemistry (PDF 69P)

This note describes
the laboratory
activities to
understand the model

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of ionic compound
close packing,
reduction-oxidation
reaction of several
metals, improve the
double salt
preparation and re-
crystallization skills as
the basic for advance
laboratory activities.

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The units of volume commonly used in this manual are cm^3 and dm^3 . The conversion between cm^3 and dm^3 is $1 \text{ dm}^3 = 1000 \text{ cm}^3$. Note that cm^3 are the same as millilitres (ml) and dm^3 are the same as litres (L). $1 \text{ cm}^3 = 1 \text{ ml}$ and $1 \text{ dm}^3 = 1 \text{ L}$. B.

Concentration

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Concentration is a measure of amount per volume. The amount can be grams, kilograms,

Guide to Chemistry
Practicals - TETEA

This manual supplies 55 experiments in 18 major subject areas, including carbohydrates, lipids, and proteins in

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biochemistry; tannins,
balsams, Medicinal
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to a critical classroom
need for material for
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Laboratory Manual by
Charles Dickson

Pharmaceutical
chemistry seeks to
examine the synthetic
methodology used in
order to produce pure
biologically active
molecules to be
formulated as
medicines. An in

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depth understanding of organic chemistry is core to the study of this subject as it plays a key role in many of the processes utilised in the formation of medicines.

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1) Collect the
following items which
you will need to carry
out the experiment.

(a) 2 100mL round

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bottom flask (b) 1
spatula (c) 1 cork ring
(d) 1 plastic bowl (e) 1
ceramic Buchner
funnel (f) 1 watch
glass (g) 1 magnetic
stirrer bar (h) 1 zip
lock bag. 2) Put on a
pair of gloves before
you start the
experiment.

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Foundations for

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Medicinal Chemistry

Chemistry

Step 1: First step involves the formation of Acetylium ion, the reaction is enhanced by proton donation by concentrated acid.

Step 2: Amino group of Para-Aminophenol dissociates as follows:

Step 3: Acetylium ion attacks the anion at the nitrogen atom to

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form Para-acetyl
Aminophenol or
Paracetamol.

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Essential Pharma

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Medicinal Chemistry-II

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chemistry ia

introduction this

course introduces the

basic concepts of

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Medicinal chemistry
including study of the
molecular and
chemical structures of
organic
pharmaceuticals
using molecular
models preparation of
some active
ingredients functional
groups reactions and
identification of
organic compounds
such as amine

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aldehyde

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Georges Simenon -

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