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*Reactions in Aqueous Solutions*  
*Chapter 4 - Reactions in Aqueous*  
*Solution: Part 1 of 6 Chapter 4 -*

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*Reactions in Aqueous Solution: Part 1  
of 8* **Chapter 4 Reactions in**

**Aqueous Solution (Sections 4.1 -  
4.4)** Chemical Reactions in Aqueous

Solutions - Part VA Chemical

Reactions in Aqueous Solutions - Part

I Chemical Reactions in Aqueous

Solutions - Part II ~~Chapter 4~~

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~~Reactions in Aqueous Solution: Part 2  
of 8 Chapter 4 Reactions in Aqueous  
Solution (Sections 4.5 - 4.6) Chapter 4  
Reactions in Aqueous Solution: Part  
7 of 8~~

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What Happens when Stuff Dissolves?  
*Aqueous Solutions, Acids, Bases and  
Salts* ~~Precipitation Reactions~~

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**Solubility Rules and Precipitation  
Reactions ChemCollective Mass of  
Silver Nitrate (Solution) Identifying  
Strong Electrolytes, Weak  
Electrolytes, and Nonelectrolytes -  
Chemistry Examples Calculating the  
Concentration of a Saturated Solution  
How to Write Complete Ionic**





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Chapter 4 - Reactions in Aqueous  
Solution: Part 3 of 8 *Chapter 4 -  
Reactions in Aqueous Solution: Part 3  
of 6* 4.1 Reactions in Aqueous  
Solutions Overview

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Lecture 6|Stoichiometry—3 Reactions  
in Aqueous Solution -1

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Reactions in Aqueous Solutions:

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Metathesis Reactions and Net Ionic  
Equations Chemical Reactions in  
Aqueous Solutions - Part VB

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Minilab An aqueous reaction is a chemical reaction that take place in water. Many important chemical reactions take place in water and many of these are associated with life. There

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Minilab 4.2: Precipitation Reactions.

(4.S.1)  $A X + B Y \rightarrow A Y + B X$ . for methathesis to occur: the formation of an insoluble product. the formation of either a weak electrolyte or a

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nonelectrolyte. the formation of a gas that escapes from solution. 4.S: Reactions in Aqueous Solution (Summary) - Chemistry ...

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Solutions Minilab Reactions in Aqueous Solutions Several types of reactions occur in water. When water is the solvent for a reaction, the reaction is said to occur in aqueous solution, which is denoted by the abbreviation (aq) following the name of a chemical species in a reaction.

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Three important types of reactions in water are precipitation, acid- base, and oxidation-reduction reactions.

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expected to describe: Metal Aqueous  
ion Action of NaOH Action of an  
excess of NaOH(aq) 3 Action of NH<sub>3</sub>  
(aq) Action of an excess

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Reactions in Aqueous Solution 10 by  
10 Matrix : Ammonia  $\text{NH}_3$ : Barium  
Chloride  $\text{BaCl}_2$ : Cobalt(II) Chloride  
 $\text{CoCl}_2$ : Copper(II) Nitrate  $\text{Cu}(\text{NO}_3)_2$ :

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Copper(II) Sulfate  $\text{CuSO}_4$ :

Hydrochloric Acid  $\text{HCl}$ : Sodium

Carbonate  $\text{Na}_2\text{CO}_3$ : Sodium

Chloride  $\text{NaCl}$ : Sodium Hydroxide

$\text{NaOH}$ : Sulfuric Acid  $\text{H}_2\text{SO}_4$  :

Ammonia  $\text{NH}_3$  : Ammonia  $\text{NH}_3$ :

Barium Chloride  $\text{BaCl}_2$  :

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## Reactions in Aqueous Solution - Department of Chemistry

EXPERIMENT: REACTIONS IN  
AQUEOUS SOLUTION:

METATHESIS REACTIONS AND NET  
IONIC EQUATIONS. solve the  
RESULTS. Show transcribed image  
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Metathesis reaction is a reaction in which two compounds react with each other to form two different compounds.

## **Solved: EXPERIMENT: REACTIONS IN AQUEOUS SOLUTION: METATHE**

...

6.11 Describe the relative reactivity of

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the halogens chlorine, bromine and iodine, as shown by their displacement reactions with halide ions in aqueous solution, and use this pattern to predict the reactions of astatine; Edexcel Combined science. Topic 6 - Groups in the periodic table. Group 7

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## **Halogens in aqueous solution and their displacement reactions**

In aqueous solution the water molecules directly linked to the metal ion form part of the first coordination sphere, also known as the first sphere of solvation or first sphere of hydration. The link between these water

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molecules and the metal ion is called  
the coordination bond : oxygen  
provides a pair of electrons to form the  
bond.

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