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Reactions Answers

Balance each of the

following reactions

and identify each

type of reaction: 1. 2

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1.  $\text{NaBr} + \text{Ca(OH)}_2 \rightarrow \text{CaBr}_2 + 2 \text{NaOH}$  double displacement  
2.  $2 \text{NH}_3 + \text{H}_2\text{SO}_4 \rightarrow (\text{NH}_4)_2\text{SO}_4$  synthesis  
3.  $\text{C}_5\text{H}_9\text{O} + 29 \text{O}_2 \rightarrow 20 \text{CO}_2 + 18 \text{H}_2\text{O}$  combustion  
4.  $3 \text{Pb} + 2 \text{H}_3\text{PO}_4 \rightarrow 3 \text{H}_2 + \text{Pb}_3(\text{PO}_4)_2$  single displacement  
5.  $\text{Li}_3\text{N} + 3 \text{NH}_4\text{NO}_3 \rightarrow 3 \text{LiNO}_3 + (\text{NH}_4)_3\text{N}$

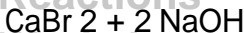
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Reaction Worksheet  
Balance the reactions  
1 to 6 and indicate  
which type of  
chemical reaction  
(synthesis,  
decomposition,  
single-displacement,  
double-displacement  
or combustion) is

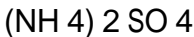
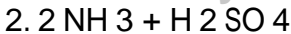
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being represented: 1.



Reaction Type :

double displacement



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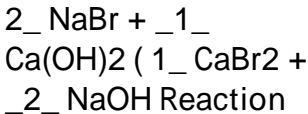
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Balance the reactions

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1 to 6. and indicate which type of chemical reaction (synthesis, decomposition, single-displacement, double-displacement or combustion) is being represented:



Reaction Type : Double replacement.

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Types Of

$\text{NH}_3 + \text{H}_2\text{SO}_4 \rightarrow$

$(\text{NH}_4)_2\text{SO}_4$  Reaction

Type : Synthesis.

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Answer Key

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Synthesis Reactions

Description: Two or  
more substances

combine to

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A 3-page worksheet  
that provides an  
overview of the 4  
basic types of



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Chemical reactions:

synthesis,  
decomposition, displ  
acement/replacemen  
t, and combustion. I

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summary reading  
exercise that acts as a  
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document to my PPT  
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Six Types of Chemical  
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Answers Balance the  
following reactions  
and indicate which of  
the six types of  
chemical reaction are  
being represented: a)  
 $2 \text{NaBr} + 1 \text{Ca(OH)}_2 \rightarrow 1$

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$\text{CaBr}_2 + 2 \text{NaOH}$  Type of reaction: double displacement  
b)  $2 \text{NH}_3 + 1 \text{H}_2\text{SO}_4 \rightarrow 1 (\text{NH}_4)_2\text{SO}_4$  Type of reaction: synthesis  
c)  $4 \text{C}_5\text{H}_9\text{O} + 27 \text{O}_2 \rightarrow 20 \text{CO}_2 + 18 \text{H}_2\text{O}$

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Balance the reactions a to e and indicate which types of ...

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some related pictures  
to complete your  
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key, chemical  
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of main things we

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will present to you  
based on the ...

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Use colored pencils  
to circle the common  
atoms or compounds  
in each equation to  
help you determine  
the type of reaction it

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illustrates. Use the code below to classify each reaction. S = Synthesis D = Decomposition SR = Single Replacement DR = Double Replacement. P + O.  
2.

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Name - Science Spot

4) 3 Types of  
*Page 24/38*



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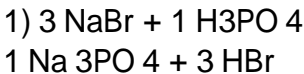
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– Solutions Balance

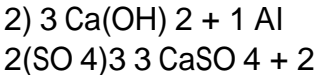
the following  
equations and

indicate the type of  
reaction taking place:



Type of reaction:

double displacement



Type of  
reaction: double

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displacement 3)  $3 \text{Mg}$

+  $1 \text{Fe}_2\text{O}_3 \rightarrow 2 \text{Fe} + 3$

$\text{MgO}$  Type of

reaction: single

displacement Types

of Reactions • There

are five types of

chemical reactions

we will use: 1. doc).

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Answers Balance the  
following reactions  
and indicate which of  
the six types of  
chemical reaction are  
being represented: a)  
 $2 \text{NaBr} + 1 \text{Ca(OH)}_2 \rightarrow 1$   
 $\text{CaBr}_2 + 2 \text{NaOH}$

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Practice Answers ...

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Worksheet  
: indicate which type  
of chemical reaction  
(synthesis,  
decomposition,  
single-displacement,  
double displacement  
or combustion) is  
being represented.

---

Classifying and

*Page 28/38*

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Balancing Chemical  
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Worksheets ...

Types of Reactions

Worksheet Balance

the following

equations and

indicate the type of

reaction taking place:

$\text{H}_3\text{PO}_4 + \text{HBr} \rightarrow \text{Na}_3\text{PO}_4 +$

$\text{NaBr} + \text{Al}_2(\text{SO}_4)_3$

Type of reaction:

$\text{CaSO}_4 +$  Type of

reaction:  $\text{Mg} + \text{C}_2\text{H}_4$

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+ PbSO<sub>4</sub> NH<sub>3</sub> + H<sub>2</sub>O +  
O<sub>2</sub> PbSO<sub>3</sub> + MgO Fe +  
Type of reaction: H<sub>2</sub>O  
CO<sub>2</sub> + Type of  
reaction: Type of  
reaction: H<sub>2</sub> Type of  
reaction: 12 SO<sub>3</sub> N<sub>2</sub> 16  
+ H<sub>2</sub>SO<sub>4</sub>

---

\*LNAN<sub>3</sub>+ PbSO<sub>4</sub>  
C<sub>3</sub>H<sub>8</sub>+ CO<sub>2</sub> 1-120 zn+  
LHCl+ +1 10. 11. 12.  
13 ...

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1 to 6. and indicate  
which type of  
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(synthesis,  
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double-displacement  
or combustion) is  
being represented:

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Results for Types Of  
Chemical Reactions  
Writing Formulas  
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Lab - Types of  
Chemical Reactions,  
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Decomposition,

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Types Of

Single Replacement,  
Double Replacement,  
Combustion,  
Balancing

equations

The purpose of this lab activity is to teach your students how to identify the types of chemical reactions. Students will be able to identify the products of each react

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Worksheet Pogil is  
perfect for the needs  
of a professional  
chemist. These  
chemical reactions  
are used to explore

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and determine the existence and effect of various substances or reactions to other substances. It is a reference sheet for the application of chemical reactions, their possible reactions, effects, advantages, and disadvantages.

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Description: General  
Equation: Drawing:  
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